

SIDE CHAIN OXIDATION OF 25-HYDROXY-[26,27-¹⁴C]VITAMIN D₃ AND
1,25-DIHYDROXY-[26,27-¹⁴C]VITAMIN D₃ IN VIVO BY CHICKENS[†]

by

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Received January 16, 1976

Radio carbon dioxide was detected in expired air of vitamin D-deficient chicks administered either 25-hydroxy-[26,27-¹⁴C]vitamin D₃ or 1,25-dihydroxy-[26,27-¹⁴C]vitamin D₃ intravenously. At 48 hours the amounts were $5.13 \pm 1.02\%$ and $11.8 \pm 3.10\%$ of the administered dose, respectively. This phenomenon, similar to the one observed in rats, represents side chain oxidation and elimination of either one or both of the 26 and 27 carbons from the vitamin D molecule. The observation that the rate of ¹⁴CO₂ evolution is maximal 4 hours after a dose of 1,25-dihydroxy-[26,27-¹⁴C]vitamin D₃ suggests that this pathway may be of importance to the function of 1,25-dihydroxyvitamin D₃.

The physiologic action of vitamin D₃ is preceded by its conversion to more polar metabolites in the liver and kidney (1-7). The active form of vitamin D₃ in intestine and bone appears to be 1,25-dihydroxyvitamin D₃ (1,25-(OH)₂D₃)¹ (6). Six and 12 hours after the administration of 1,25-(OH)₂-[26,27-¹⁴C]D₃ to the rat, 1,25-(OH)₂D₃ is the major metabolite present in these tissues (8, 9). However, a substantial amount of radioactivity was present in the aqueous phase suggesting unknown metabolism. To investigate this matter further 25-hydroxy-[26,27-¹⁴C]vitamin D₃ (25-OH-D₃) was synthesized using established methods (10). 25-OH-D₃ was then converted to 1,25-(OH)₂D₃ enzymatically using chick kidney homogenates (8).

[†]This work was supported by PHS grant No. AM-14881 and the Steenbock Fund of the Wisconsin Alumni Research Foundation.

¹Abbreviations: 1,25-(OH)₂D₃, 1,25-dihydroxyvitamin D₃; 25-OH-D₃, 25-hydroxyvitamin D₃.

Table 1. Accumulative $^{14}\text{CO}_2$ Production After the Injection of
 $25\text{-OH-[26,27-}^{14}\text{C]D}_3$ and $1,25\text{-(OH)}_2\text{-[26,27-}^{14}\text{C]D}_3$
into Vitamin D-Deficient Chickens

25-OH-D ₃		1,25-(OH) ₂ D ₃	
Time (hours)	% Dose	Time (hours)	% Dose
12	2.55 ± 0.50	12	6.62 ± 2.85
24	4.22 ± 0.91	24	10.56 ± 3.12
48	5.13 ± 1.02	48	11.18 ± 3.10

Rachitic chicks were given either 650 pmoles $25\text{-OH-[26,27-}^{14}\text{C]D}_3$ or 325 pmoles $1,25\text{-(OH)}_2\text{-[26,27-}^{14}\text{C]D}_3$ intravenously in 0.05 ml ethanol. There were 3-4 chicks in each group and the data are expressed as % of dose ± S.D.

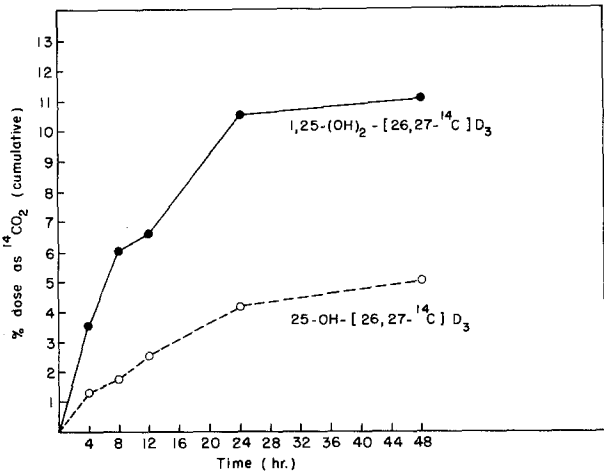


Figure 1. $^{14}\text{CO}_2$ appearance in the expired air of rachitic chicks given either 650 pmoles $25\text{-OH-[26,27-}^{14}\text{C]D}_3$ or 325 pmoles $1,25\text{-(OH)}_2\text{-[26,27-}^{14}\text{C]D}_3$ intravenously.

White leghorn chickens (Northern Hatcheries, Beaver Dam, Wis.) were raised on a vitamin D-deficient diet (11) for a period of 3 weeks. Two groups of 4 chicks were used. Animals in the first group were injected intravenously with 650 pmoles of 25-OH-[26,27- ^{14}C]D₃ and the second group with 325 pmoles of 1,25-(OH)₂-[26,27- ^{14}C]D₃ each dissolved in 0.05 ml ethanol. The animals were immediately placed in metabolism cages and all the CO₂ ($^{12}\text{CO}_2$ and $^{14}\text{CO}_2$) expired was trapped in ethanolamine-methyl cellulose (1:2) solution (12). This solution was changed periodically at 4, 8, 12, 24, and 48 hours. Aliquots of this were taken and counted in a liquid scintillation counter (12).

$^{14}\text{CO}_2$ was detected in expired air in both groups of chickens. The amounts formed are shown in Table 1 and Figure 1. It is evident that the rate of formation of $^{14}\text{CO}_2$ is maximal in the first 4-8 hours after the administration of 1,25-(OH)₂-[26,27- ^{14}C]D₃.

This phenomenon is similar to the one observed in rats (13, 14). In rats nephrectomy prevented $^{14}\text{CO}_2$ formation from 1,25-(OH)₂-[26,27- ^{14}C]D₃ and conditions suppressing 25-OH-D₃-1-hydroxylase in the kidney (e.g. vitamin D₃ repletion) also suppress $^{14}\text{CO}_2$ evolution.

$^{14}\text{CO}_2$ in both groups of animals is formed from 25-OH-[26,27- ^{14}C]D₃ and 1,25-(OH)₂-[26,27- ^{14}C]D₃ as a result of loss of either one or both carbons at positions 26 and 27. The fact that $^{14}\text{CO}_2$ formation is maximal at a time when 1,25-(OH)₂D₃ function is most marked suggests that the pathway may be of importance in 1,25-(OH)₂D₃ action. It is also possible that this represents a degradative pathway after 1,25-(OH)₂D₃ has stimulated calcium and phosphorus transport in the intestine and calcium mobilization from bone.

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